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Key indicators

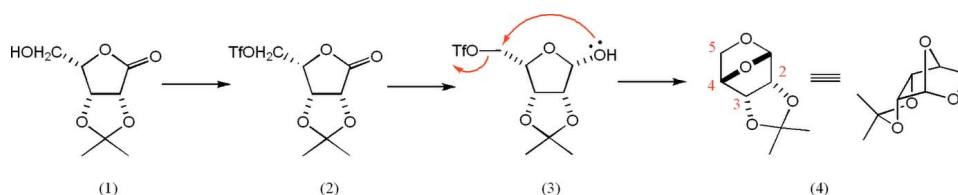
Single-crystal X-ray study
T = 150 K
Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$
R factor = 0.031
wR factor = 0.031
Data-to-parameter ratio = 7.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.1,5-Anhydro-2,3-O-isopropylidene-L-lyxofuranose [(1*S*,4*S*,5*S*,6*R*)-5,6-O-isopropylidene-2,7-dioxabicyclo[2.2.1]heptane-5,6-diol]

The tricyclic title compound, $\text{C}_8\text{H}_{12}\text{O}_4$, was formed in the reduction by diisobutylaluminium hydride of a 5-*O*-trifluoromethanesulfonyl lactone and is likely to be useful as a chiral intermediate for the synthesis of bioactive compounds. The absolute configuration was determined by the use of 2,3-*O*-isopropylidene-L-lyxono-1,4-lactone as the starting material.

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Comment

Branched sugar lactones may be useful as versatile intermediates in the synthesis of novel branched biologically important molecules and biopolymeric materials (Asano *et al.*, 2000; Ichikawa & Igarashi, 1995; Ichikawa *et al.*, 1998; Soengas *et al.*, 2005; Hotchkiss *et al.*, 2004, 2006). The Ho crossed-aldol reaction (Ho, 1979, 1985) of lactols with formaldehyde is one of the most powerful strategies for the synthesis of these compounds. In the course of the synthesis of a branched sugar lactone by a Ho procedure (Simone *et al.*, 2005), the trifluoromethanesulfonate, (2), derived from the 2,3-*O*-isopropylidene-L-lyxono-1,4-lactone, (1) (Simone *et al.*, 2005), was treated with diisobutylaluminium hydride (DIBAL-H). A rationalization for the formation of the title compound, (4), is that initial reduction afforded the β -lyxose, (3), which spontaneously cyclized to (4) by $\text{S}_{\text{N}}2$ displacement of the trifluoromethanesulfonate. The value of such highly oxygenated bicyclic intermediates as (4) is well established (Cossy *et al.*, 1995; Pechy *et al.*, 1993), and the unexpected formation of (4) may provide another valuable chiron.



The structure of (4) has been determined by X-ray diffraction (Fig. 1), which showed that it is formed by a six-membered ring fused to a five-membered ring (the acetonide protecting group). The five-membered ring adopts an envelope conformation. The six-membered ring is a $\text{B}_{2,5}$ -type conformationally constrained structure, due to the existence of the oxygen bridge linking the 1- and 4-positions.

Experimental

1,5-Anhydro-2,3-*O*-isopropylidene-L-lyxofuranose, (4), was obtained in 28% yield upon overnight reduction of the trifluoromethane-

sulfonate (2) with DIBAL-H in tetrahydrofuran at low concentration (16 mg ml⁻¹). The title material, (4), isolated as a crystalline but volatile product, was recrystallized *via* solvent evaporation (ethyl acetate–cyclohexane) (R_f 0.57; m.p. 338–339 K). HRMS (FI⁺), found: 172.0732 [M]⁺; C₈H₁₂O₄ requires: 172.0736; [α]_D²¹ 106.8 (*c*, 0.72 in methanol); IR (thin film, ν_{\max} , cm⁻¹): 2992, 2955, 2906 (*s*, C–H), 1378 (–O–CO–CH), 1291 (C–O); ¹H NMR (CDCl₃, 400 MHz): δ 1.34, 1.60 [2 × 3H, 2 × *s*, C(CH₃)₂], 3.55 (1H, *ddd*, $J_{H5,H5'} = 6.6$ Hz, $J_{H5,H4} = 3.5$ Hz, $J_{H5,H3} = 1.3$ Hz, H5), 4.31 (1H, *a-d*, $J = 6.8$ Hz, H5'), 4.45 (1H, *a-dd*, $J = 8.2$ and 2.3 Hz, H2), 4.64 (1H, *dd*, $J_{H3,H2} = 8.1$ Hz, $J_{H3,H4} = 4.6$ Hz, H3), 4.72–4.77 (1H, *m*, H4), 5.45 (1H, *d*, $J_{H1,H2} = 2.2$ Hz, H1); ¹³C NMR (CDCl₃, 100 MHz): δ 25.6, 26.1 [C(CH₃)₂], 63.7 (C5), 76.5 (C3), 78.8 (C4), 81.6 (C2), 100.1 (C1), 119.7 [C(CH₃)₂].

Crystal data

C ₈ H ₁₂ O ₄	$Z = 4$
$M_r = 172.18$	$D_x = 1.370$ Mg m ⁻³
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation
$a = 8.1279$ (3) Å	$\mu = 0.11$ mm ⁻¹
$b = 9.4993$ (4) Å	$T = 150$ K
$c = 10.8126$ (4) Å	Fragment, colourless
$V = 834.83$ (6) Å ³	0.38 × 0.32 × 0.14 mm

Data collection

Nonius KappaCCD area-detector diffractometer	5825 measured reflections
ω scans	1111 independent reflections
Absorption correction: multi-scan (DENZO and SCALEPACK; Otwinowski & Minor, 1997)	834 reflections with $I > 3\sigma(I)$
$T_{\min} = 0.96$, $T_{\max} = 0.98$	$R_{\text{int}} = 0.037$
	$\theta_{\text{max}} = 27.4^\circ$

Refinement

Refinement on F	0.0219 $T_2(x)$ where T_i are Chebyshev polynomials and $x = F_c/F_{\text{max}}$ (Prince, 1982; Watkin, 1994)
$R[F^2 > 2\sigma(F^2)] = 0.031$	
$wR(F^2) = 0.031$	
$S = 1.15$	
834 reflections	$(\Delta/\sigma)_{\text{max}} = 0.005$
109 parameters	$\Delta\rho_{\text{max}} = 0.14$ e Å ⁻³
H-atom parameters constrained	$\Delta\rho_{\text{min}} = -0.14$ e Å ⁻³
$w = [1 - (F_o - F_c)^2/36\sigma^2(F)]^2 / [0.275T_0(x) + 0.0396T_1(x) + \dots]$	

A large single-crystal was cut to give a small fragment. This was mounted on a glass fibre using perfluoropolyether oil and cooled rapidly to 150 K in a stream of cold N₂ using an Oxford Cryosystems CRYOSTREAM unit. H atoms were positioned geometrically after each cycle of refinement, with C–H = 1.00 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: COLLECT (Nonius, 2000); cell refinement: DENZO (Otwinowski & Minor, 1997); data reduction: DENZO and SCALEPACK (Otwinowski & Minor, 1997); program(s) used to solve structure: SIR92 (Altomare *et al.*, 1994); program(s) used to refine structure: CRYSTALS (Betteridge *et al.*, 2003); molecular graphics: CAMERON (Watkin *et al.*, 1996); software used to prepare material for publication: CRYSTALS.

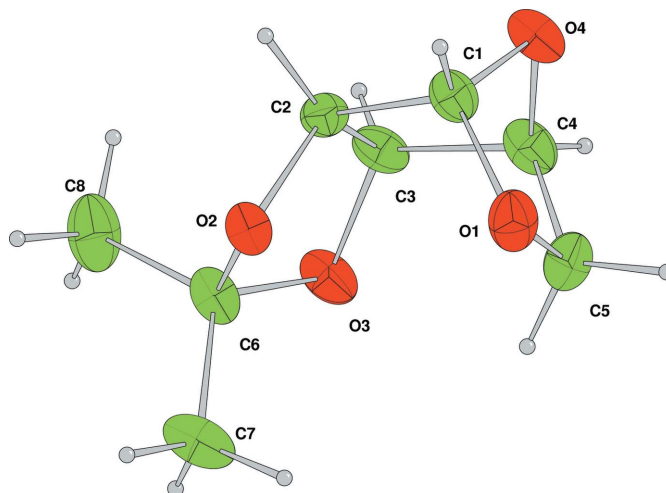


Figure 1

The molecular structure of the title compound, with displacement ellipsoids drawn at the 50% probability level. H atoms are shown as spheres of arbitrary radius.

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